product on silica gel plates afforded a 17% yield of racemic cryptosporiopsin<sup>7</sup>, whose identity was established by comparison of IR- (CCl<sub>4</sub>), UV-, NMR- and mass-spectra, as well as TLC behavior, with those of natural cryptosporiopsin.

The synthetic racemic antibiotic was assayed for its activity against sporangial germination of *Phythophthora infestans*  $^8$  in aqueous solution. Germination was almost completely prevented at a concentration of 12.5  $\mu$ g/ml. Natural (dextrorotatory) cryptosporiopsin showed about the same degree of inhibition at 6.25  $\mu$ g/ml. (Control; 70% germination.) These results suggest that the dextrorotatory enantiomer alone is responsible for the observed inhibition of sporangial germination.

Résumé. L'antibiotique cryptosporiopsine, produit métabolique de Sporormia affinis ainsi que d'une espèce de Cryptosporiopsis a été synthétisé à partir de la dihydrocryptosporiopsine synthétique.

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- $^7$  The yield was 20% based on unrecovered starting material.
- 8 M. A. STILLWELL and W. A. HODGSON, Can. J. Microbiol. 14, 807 (1968). We thank Mr. M. A. STILLWELL for this determination.

## Terpenoid Alkaloids from Murraya koenigii Spreng. IV1 Structure and Synthesis of Mahanimbinine2

Murraya koenigii Spreng. has proved to be a rich source of terpenoid carbazole alkaloids. Up to date 12 of these have already been reported  $^{8-11}$ . The present communication describes the structure and synthesis of one more base – a congener of mahanimbine (I) from the leaves of this plant.

apart from its partial racemic nature, with the natural product. Since the synthesis of mahanimbine has already been reported<sup>1</sup>, this constitutes the total synthesis of the new base. The partial racemization of this base as well as of some other members in this series will be discussed in a subsequent communication.

Reagents: 1, m-Chloroperbenzoic acid. 2, LiAlH,

The alkaloid named mahanimbinine,  $C_{23}H_{27}NO_2$  (M<sup>+</sup>, 349), mp 179°;  $\nu_{max}$  (CHCl<sub>3</sub>) 3580 (OH), 3450 (NH), 1630 and 1600 cm<sup>-1</sup> (unsaturation and aromatic system) had an UV-spectrum,  $\lambda_{max}$  (EtOH), 238, 288, 329, 344, and 359 nm (log  $\varepsilon$  4.64, 4.61, 3.83, 3.87, and 3.82 respectively). The NMR-spectrum (CDCl<sub>3</sub>) showed the following signals:

$$\tau$$
 8.80, s, 6,  $-O-C < \frac{CH_3}{CH_2}$ ; 8.59, s, 3,  $-O-C-CH_3$ ; 7.67,

s, 3, ar. CH<sub>3</sub>; 8.14–8.65, m, 6, methylene protons; 4.42, d (J 10 Hz), 1, olefinic H; 3.40, d (J 10 Hz), 1, benzylic methine H; 2.35, s, 1, 4-H; 2.07, m, 1, 5-H. There were 3 more aromatic protons in the region 2.55–2.94  $\tau$ .

The mass spectrum of the base showed, apart from the molecular ion peak,  $M^+$  349, abundant ions at m/e 334, 331, 330, 316, 276, 275, 261, 260, 249, 248 (base peak), 247, 234, 218, 210, 204, and 180. The combined data and particularly the correspondence of the base peak with that obtained from mahanimbine  $^{10}$  (I), the absence of olefinic protons in the side-chain, the presence of a -OH group in the IR-spectrum and the  $M^+$  at m/e 349 (18 units higher than that of mahanimbine) led to constitution (III) for mahanimbinine.

This was confirmed by its synthesis as follows: (+)-mahanimbine (I) stirred 3 h at room-temperature with m-chloroperbenzoic acid in dry ether gave the epoxy compound (II) as the major product. Reduction of (II) with LiAlH<sub>4</sub> followed by purification of the product on silica-gel column gave a compound in 30% yield, mp 148° (benzene) which on the basis of elemental analysis, TLC, UV-, IR- and NMR-spectra, was identical,

Zusammenfassung. Die Struktur (III) des Mahanimbinins, eines Verwandten des Mahanimbins aus den Blättern von Murraya koenigii Spreng., ist spektroskopisch und durch Synthese aufgeklärt worden.

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Central Drug Research Institute, Lucknow (India), 20 April 1970.

- <sup>1</sup> Part III. S. P. KUREEL, R. S. KAPIL and S. P. POPLI, Chem. Commun. (1969), 1120.
- $^2$  Communication No. 1505 from the Central Drug Research Institute, Lucknow.
- <sup>3</sup> D. P. CHAKRABORTY and B. K. CHOWDHURY, J. org. Chem. *33*, 1265 (1968)
- <sup>4</sup> D. P. CHARABORTY and K. C. Das, Chem. Commun. (1968), 967.
- <sup>5</sup> N. S. NARASIMHAN, M. V. PARADKAR and V. P. CHITGUPPI, Tetrahedron Lett. (1968), 5501.
- <sup>6</sup> B. K. CHOWDHURY and D. P. CHAKRABORTY, Chem. Ind. (1969), 549.
- <sup>7</sup> D. P. CHAKRABORTY, B. K. BARMAN and P. K. Bose, Sci. Cult. 30, 445 (1964). – N. L. DUTTA and C. QUASIM, Ind. J. Chem. 7, 307 (1969).
- <sup>8</sup> S. P. Kureel, R. S. Kapil and S. P. Popli, Experientia 25, 790 (1969).
- 9 S. P. KUREEL, R. S. KAPIL and S. P. POPLI, Tetrahedron Lett. (1969), 3857.
- S. P. KUREEL, R. S. KAPIL and S. P. POPLI, unpublished work.
  N. L. DUTTA C. QUASIM and M. S. WADIA Ind. J. Chem. 7, 1061 (1969).